Honors Earth Science: Chapter 4 - Weathering, Soil, and Mass Wasting

- I. Earth's external processes include
 - A. Weathering— the disintegration and decomposition of material at or near the surface
 - B. Mass wasting— the transfer of rock material downslope under the influence of gravity

Water, wind, gravity, glacier

C. Erosion— the incorporation and transportation of material by a mobile agent, usually water, wind, or ice

II. Weathering

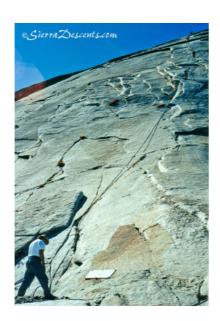
- A. Two kinds of weathering
 - Mechanical weathering

No chemical change in the rock
Big rock ----> Little rock

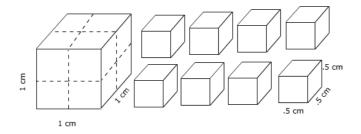
- Breaking of rocks into smaller pieces
- Four processes
 - Frost wedging water freezes expanding about 10%
 - Unloading exfoliation
 - Thermal expansion
 - Biological activity Burrowing animals Root Wedging
- 2. Chemical weathering
 - a. Alters the internal structures of minerals by removing or adding elements
 - Most important agent is water



- 1. Oxygen dissolved in water oxidizes materials
- Carbon dioxide (CO2) dissolved in water forms carbonic acid and alters the material



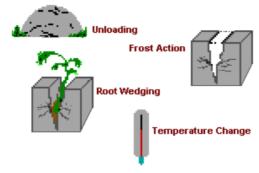




Volume = 1 cm³ Total Surface Area = 6 cm⁹

Volume = 1 cm³ Total Surface Area = 12 cm⁹

Mechanical Weathering Includes:



HALF OF THE WORLD'S PEOPLE LIVE ON LESS THAN $\$2\,$ A DAY...



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... AND WE WONDER WHY THE WORLD HAS PROBLEMS?

 $2KAlSi_3O_8 \ + \ 2(H^+ + HCO_3^-) \ + H_2O ----- > Al_2Si_2O_5(OH)_4 + 2K^+ \ + \ 2\ HCO_3^- \ + \ 4SiO_2$

Potassium Feldspar Carbonic Acid

Clay Mineral Potassium Ion

Potassium Ion Bica Solution

Bicarbonate Silica

c. Weathering of granite

Water

Glue for Sedimentary Rocks

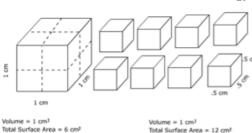
- 1. Weathering of Potassium (K)-feldspar produces
 - a. Clay minerals
 - b. Soluble salt (Potassium bicarbonate)
 - Silica in solution
- 2. Quartz remains substantially unaltered
- 3. Weathering of silicate minerals produces



- a. Insoluble iron oxides
- b. Clay minerals Sediments
- c. Soluble sodium, calcium, potassium and magnesium products
- d. Can also produce physical changes by spheroidal weathering.

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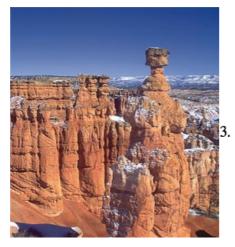
- B. Rates of weathering how fast or how slow
 - Advanced mechanical weathering aids chemical weathering by increasing the surface area
 - 2. Other important factors are
 - a. Rock characteristics
 - Mineral composition and solubility



- a. Marble (calcite) readily dissolves in weakly acidic solutions
- Silicate minerals weather in the same order as their order of crystallization

High temp minerals -- prone to weathering more quickly Physical features such as joints

- b. Climate
- Climate 30 year average of daily weather (temperature, precipitation)
 - 1. Temperature and moisture are the most crucial factors
 - Chemical weathering is most effective in areas of warm temperatures and abundant moisture



warm temperatures and abundant moisture

Differential weathering

- a. Caused by variations in composition
- Creates unusual and spectacular rock formations and landforms.

25% Water

Humus

- II. Soil
 - A. An interface in the Earth system
- B. A combination of mineral matter, water, and air—that portion of the regolith (rock and mineral fragments) that supports the growth of plants.

 25% Air

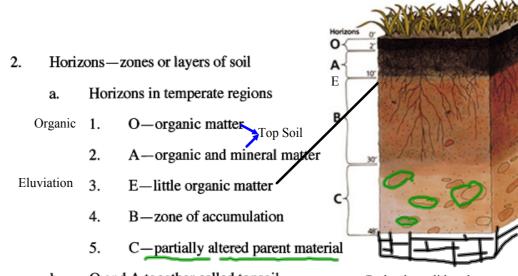
 45% Regolith
 - C. Soil texture and structure
 - 1. Texture
 - a. Refers to the proportions of different particle sizes
 - Sand (large size)
 - 2. Silt Regolith
 - Clay (small size)
 - b. Loam is best suited for plant life

Structure

- a. Soil particles clump together to give a soil its structure
- b. Four basic soil structures

 Peds loose soil particles
 - Platy
 - 2. Prismatic
 - 3. Blocky
 - 4. Spheroidal

- C. Controls of soil formation
 - 1. Parent material Rock from which the soil weathered
 - a. Residual soil—parent material is the bedrock
 - b. Transported soil—parent material has been carried from elsewhere and deposited
 - 2. Time
 - a. Important in all geologic processes
 - b. Amount of time to evolve varies for different soils
 - 3. Climate *** Most Important**
 30 year average of daily weather
 - 4. Plants and animals
 - a. Organisms influence the soil's physical and chemical properties
 - Furnish organic matter to soil
 - 5. Slope
 - a. Angle
 - Steep slope—often poor soils
 - 2. Optimum is a flat-to-undulating upland surface
 - B. Orientation (direction the slope is facing) influences
 - 1. Soil temperature
 - Moisture
- D. Soil Profile
 - 1. Soil forming processes operate from the surface downward



b. O and A together called topsoil

Bedrock - solid rock

c. O, A, E, and B together called solum, or "true soil"

E. Soil types

- 1. Hundreds of soil types worldwide 8 types in Cumberland County, PA
- 2. Three very generic types
 - a. Pedalfer

Fe

- 1. Accumulation of iron oxides and Al-rich clays in the B horizon
- 2. Best developed under forest vegetation
- b. Pedocal
 - 1. Accumulate calcium carbonate
 - Associated with drier grasslands
- c. Laterite

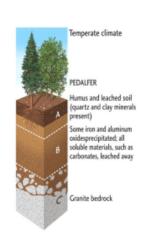
High temperatures

- 1. Hot, wet, tropical climates accelerate chemical reactions
- 2. Intense chemical weathering

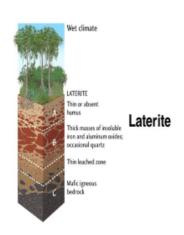
 Very limited thin Topsoil



Pedocal



Pedalfer







F. Soil Erosion

- 1. Recycling of Earth materials
- 2. Natural rates of erosion depend on
 - a. Soil characteristics
 - b. Climate
 - c. Slope
 - d. Type of vegetation
- 3. Soil erosion and sedimentation can cause
 - a. Reservoirs to fill with sediment
 - b. Contamination by pesticides and fertilizers

IV. Weathering creates ore deposits

- Process called secondary enrichment
 - 1. Concentrates metals into economical deposits
 - 2. Two ways of enrichment
 - a. Removing undesired material from the decomposing rock, leaving the desired elements behind
 - Desired elements are carried to lower zones and deposited

B. Examples

- 1. Bauxite, the principal ore of aluminum
- 2. Many copper and silver deposits

V. Mass Wasting

- A. The downslope movement of rock, regolith, and soil under the direct influence of gravity
- B. Gravity is the controlling force
- C. Important triggering factors are
 - 1. Saturation of the material with water
 - a. Destroys particle cohesion
 - b. Water adds weight
 - Oversteepening of slopes
 - a. Unconsolidated granular particles assume a stable slope called the angle of repose.
 - b. Stable slope angle is different for various materials
 - c. Oversteepened slopes are unstable
 - 3. Removal of anchoring vegetation
 - 4. Ground vibrations from earthquakes
- D. Types of mass wasting processes

4.

1. Generally each type is defined by

Rock

1. Debris
2. Mud What is moved?
How it moves?
How fast it moves?

b. The movement of the material

- 1. Fall (free-fall of pieces)
- 2. Slide (material moves along a surface)
- 3. Flow (material moves as a viscous fluid)
- c. The velocity or rate of the movement
 - 1. Fast
 - 2. Slow
- 2. Forms of mass wasting
 - a. Slump
 - 1. Rapid
 - 2. Movement along a curved surface
 - 3. Along oversteepened slopes
 - b. Rockslide
 - Rapid
 - 2. Blocks of bedrock move down a slope
 - c. Debris / Mudflow
 - 1. Rapid

Mud: Soil or ash

- 2. Flow of debris with water
- 3. Often confined to channels
- 4. Serious problem in dry areas with heavy rains
- 5. Debris flows compsed mostly of volcanic materials on the flanks of volcanoes are called lahars.

d. Earthflow

Avalanche - snow moving downslope

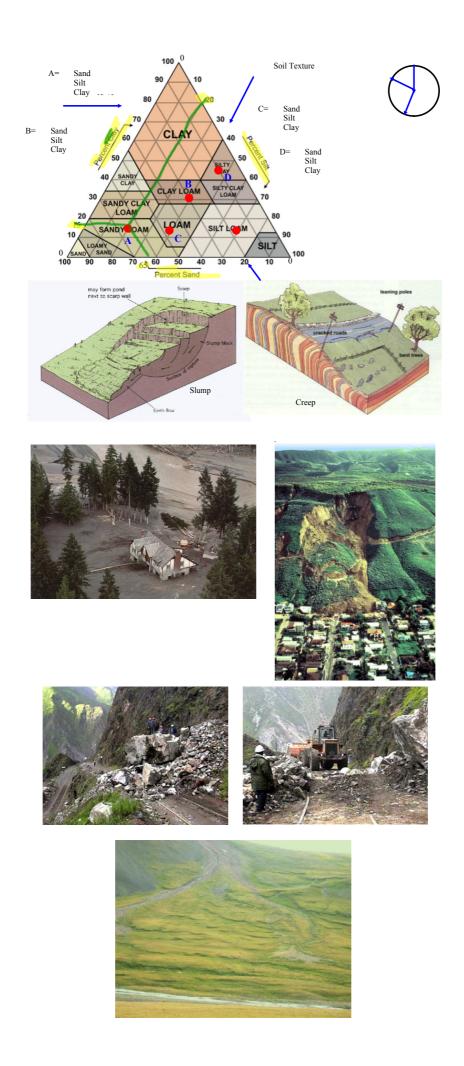
- 1. Rapid
- 2. On hillsides in humid regions
- 3. Water saturates the soil
- Liquefaction- a special type of earthflow sometimes associated with earthquakes.

Slow

e. Creep

Most Common and Widespread Insurance - most expensive

- Slow movement of soil and regolith downhill
- 2. Causes fences and utility poles to tilt
- Solifluction
 - 1. Slow
 - 2. In areas underlain by permafrost
 - 3. Upper (active) soil layer becomes saturated and slowly flows over a frozen surface below.



The level of carbon dioxide in the atmosphere has been increasing for more than a century. Should this increase tend to accelerate or slow down the rate of chemical weathering of Earth's surface rocks? What exactly is the link between chemical weathering and carbon dioxide?

Describe some possible sources of this increasing level of carbon dioxide.

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