

ARC Week at Glance – Meena (S2, W4)

Topic: Types of Energy & Transformations

Course: Phy.Sci

Grade: 9-12

Dates: January 27-31

	Learning Target (I am learning ...)	Criteria for Success (I can...)	Activation/ Instruction	Collaboration/ Guided Practice	Independent Learning/ Assessment
			<i>(Include at least one/two formatives*in any part of the lesson as needed)</i>		
<i>Monday</i>	<i>I am learning about energy and transformations.</i>	<i>I can... -Name the different types of energy · --Describe how energy flows from one form to another in consequential order</i>	<i>Do Now: MCQ'S and justification</i>	<i>Energy transformations brainstorm—group work. Presentation of their examples/ gallery walk.</i>	<i>Exit ticket: write one thing that you understood in today's assignment.</i>
<i>Tuesday</i>	<i>I am learning about energy and transformations</i>	<i>I can... · Explain how molecular motion relates to thermal energy changes · Compare and contrast conduction, convection, and radiation</i>	<i>Do Now: Review questions and explanation on types of energy and transformations.</i>	<i>Forms of thermal transfer – differentiating conduction, convection and radiation</i>	<i>Assessment check—energy transformations.</i>
<i>Wednesday</i>	<i>I am learning about energy and transformations</i>	<i>I can... -- Determine whether a form of thermal energy transformation is conduction, convection, or radiation ·</i>	<i>Do now: conduction, convection and radiation –previous day concept.</i>	<i>Conduction, convection, radiation classification worksheet.</i>	<i>TOTD: What did you understand from today's lesson?</i>
<i>Thursday</i>	<i>I am learning about energy and transformations</i>	<i>I can... -- Explain how thermal energy is transferred through heat.</i>	<i>Do now: MCQ's and justification</i>	<i>Introduce the concept specific heat. Specific heat data--Read and analyze the appropriate material for different applications.</i>	<i>TOTD: Pick correct material for specific applications.</i>
<i>Friday</i>	<i>I am learning about energy and transformations</i>	<i>I can... --Test different conductors and insulators to prove specific heat data · Read and analyze specific heat data</i>	<i>Do Now: Specific heat data – review and previous day data.</i>	<i>Specific heat data practice 1</i>	<i>Specific heat data practice 2</i>

ARC Week at Glance – Meena (S2, W3)

Topic: net ionic equation Course: AP Chemistry

Grade: 9-12

Dates: January 27-31

	Learning Target (I am learning ...)	Criteria for Success (I can...)	Activation/ Instruction	Collaboration/ Guided Practice	Independent Learning/ Assessment
			(Include at least one/two formatives*in any part of the lesson as needed)		
Monday	I am learning about chemical reactions.	I can --Write net ionic equations for double and single replacement reactions that produce precipitates, gases, or molecular compounds.	Bell ringer: Review writing formulas, balancing etc	Power point presentation to help students predict the meaning of the term net ionic equation. --stress the importance of physical states. --help students understand solubility rules thereby identify the precipitate in a double replacement reaction	Net ionic equation worksheet
Tuesday	I am learning about chemical reactions.	I can --Use the solubility rules to predict precipitate formation	Bell Ringer: MCQ'S with justification	Students will work in small group/pairs to complete an assignment to write the molecular equation, total ionic and net ionic equation. --discuss the strategy to identify the spectator ions.	Net ionic equation—college board videos
Wednesday	I am learning about chemical reactions.	I can.. Identify what species are really present in an aqueous solution. --Write net ionic equations for reactions that occur in aqueous solution.	Do Now: Review solubility rules.	Net ionic equation practice #1	Ed puzzle video—net ionic equation and assessment

<i>Thursday</i>	<i>I am learning about chemical reactions.</i>	<i>I can.. --Identify possible products: insoluble ionic compound, water, weak electrolyte --Write net ionic equations for reactions that occur in aqueous solution.</i>	<i>Do Now: Review molecular, ionic equation and spectator ions.</i>	<i>Net ionic equation practice #2 and Ed puzzle video</i>	<i>College board videos.</i>
<i>Friday</i>	<i>I am learning about chemical reactions.</i>	<i>I can ... Demonstrate my understanding on the concept of net ionic equation.</i>	<i>Review for the test.</i>	<i>Unit test: net ionic equation.</i>	<i>Go over the answers. Discuss the plan for retest.</i>

Additional Info:

Literacy Task

Minor Grade

Major Grade

Course materials and resources are available in Canvas